

# Spectral values

As we saw in the first chapter on light + eye = color,







The eye collects information about electromagnetic frequencies, and the translation of these radiations by the brain is the sensation of color.

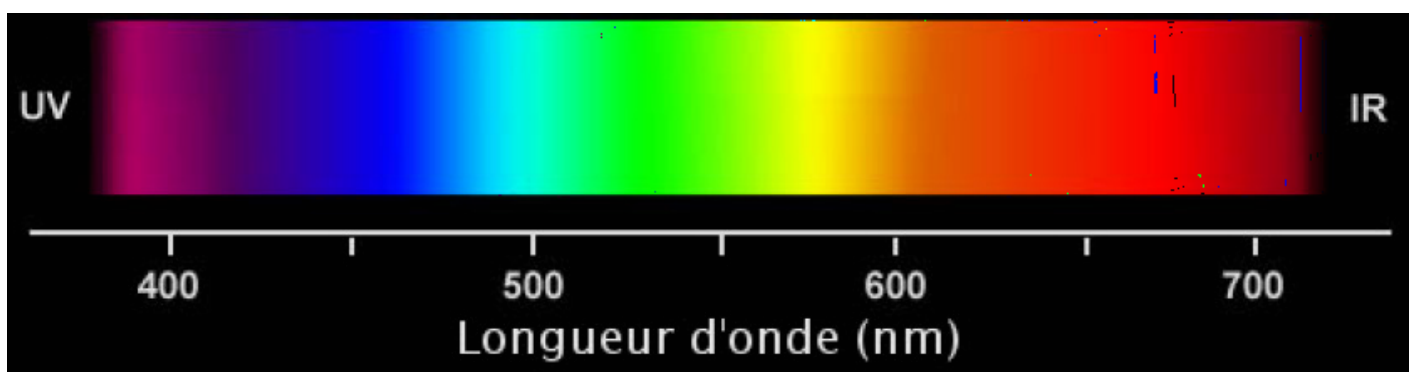
Let's take a closer look at the visible spectrum.

## Monochromatic color

One qualifies as monochromatic (from the Greek *mono-*, *one* and *chromos*, *color*) a light whose color is formed only of a frequency or, of a very narrow band of frequency at the level of its spectrum.

A source monochromatic emits electromagnetic radiation at a length of wave accurately. For example, laser radiation has only one component (one wavelength), so it has monochromatic light emission. The visible spectrum allows only a certain number of monochromatic colors

Couleur		longueur d'onde (1 nm – 10.9 m)	Fréquence (103 GHz – 1012 Hz)
violet		380 à 450 nm	725
bleu		450 à 490 nm	640
vert		490 à 570 nm	565
jaune		570 à 585 nm	520
orange		585 à 620 nm	500
rouge		620 à 670 nm	465



## Polychromatic Color

Why does the color pink not appear in the visible light spectrum?

Because a color is by definition associated with a wavelength of electromagnetic radiation (or a very restricted range).

Pink is not associated with a wavelength, so it is not a color. (Just like brown, black, white). It is associated with several wavelengths (it has a wide spectral range).

If we perceive “pink” well, it is because our eyes are endowed with three types of receptors sensitive to blue, red and green.

With the correct mixture (in certain proportions) of spectral components, you simultaneously stimulate the three types of receptors, and our brain carries out an additive synthesis of these three simultaneous stimuli. This results in a “well defined” sensation which is that of pink!

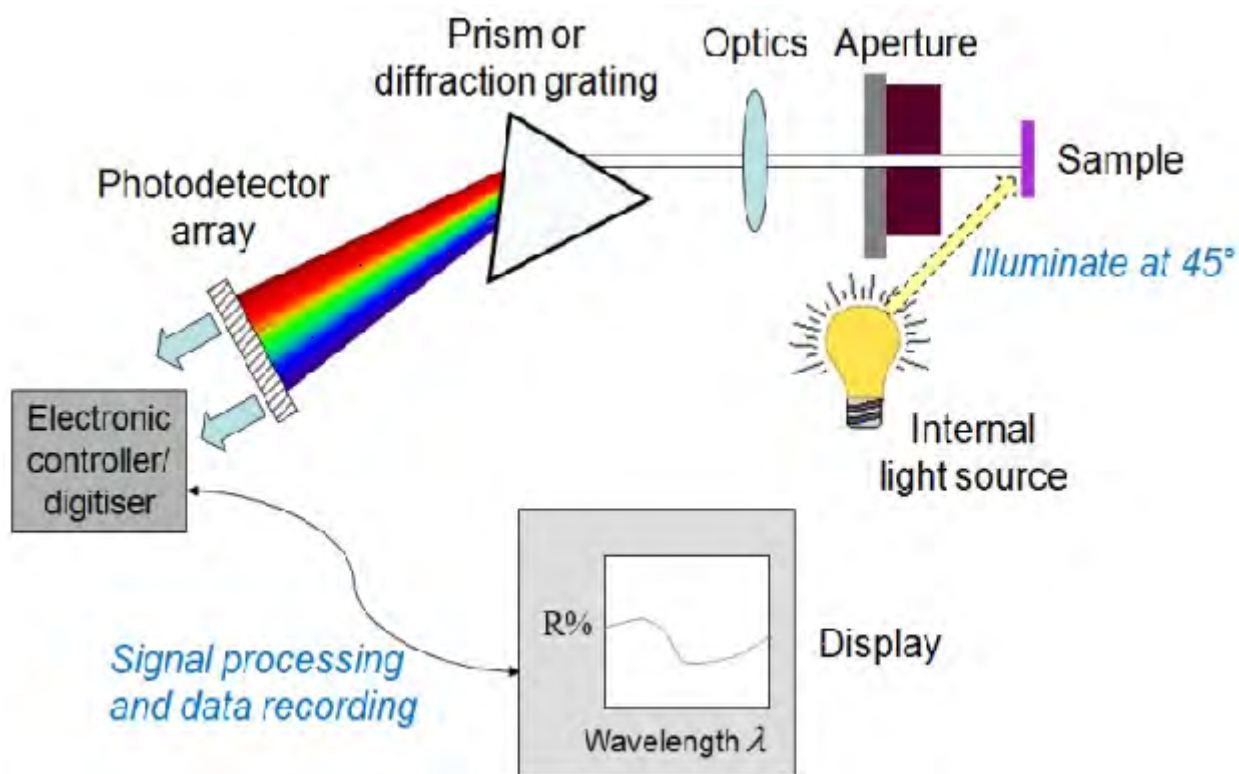
Just like white which is the sum of the wavelengths of the visible spectrum.

## The spectrophotometer

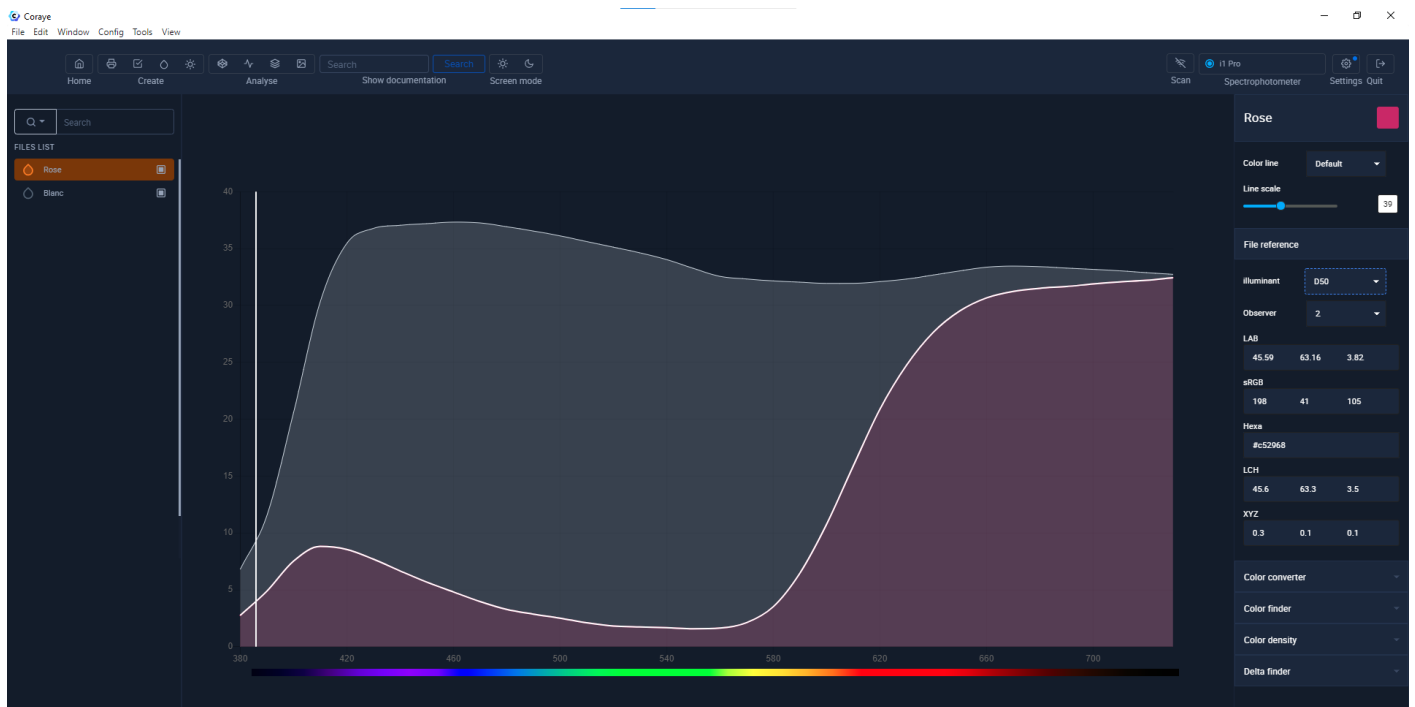
As we saw in the chapter on spectrophotometers, the measured sample sends back waves that it has not absorbed back to the spectrophotometer.

The prism system will break down these waves to redirect them to a photosensitive sensor.

The received signal, once amplified, will display the reflectance curve (spectral curve) corresponding to the measured "color".



Result of the measurement of a "Pink" sample and a "White" sample



Result of the measurement of a "Pink" sample and a "White" sample

**The color is saved as a file containing the values of the spectral curve. The XYZ and Lab values were calculated by an algorithm which converts the spectral curve into a color seen by the human eye.**

```

1 SPECT
2
3 DESCRIPTOR "Coraye power/reflectance information"
4 ORIGINATOR "Lionel Wetteren <lionel@coraye.com>"
5 MANUFACTURER "Coraye"
6 MEAS TYPE "REFLECTIVE"
7 SPECTRAL_BANDS "36"
8 SPECTRAL_START_NM "380.000000"
9 SPECTRAL_END_NM "730.000000"
10 SPECTRAL_NORM "100.000000"
11
12 NUMBER_OF_FIELDS 42
13
14 BEGIN_DATA_FORMAT
15 LAB_L LAB_A LAB_B XYZ_X XYZ_Y XYZ_Z SPECTRAL_380 SPECTRAL_390 SPECTRAL_400 SPECTRAL_410 SPECTRAL_420 SPECTRAL_430 SPECTRAL_440 SPECTRAL_450 SPECTRAL_460 SPECTRAL_470 SPECTRAL_480 SPECTRAL_
16 END_DATA_FORMAT
17
18 NUMBER_OF_SETS 1
19
20 BEGIN_DATA
21 45.5879938084934 63.16047341622966 3.8191696730270452 0.27375542433953576 0.14966291668042406 0.1106361019198491 7.06273 12.520800000000000 19.2975 22.618299999999999 21.829000000000000 19.7331 17.0972 14.563699999999999 1
22
23 END_DATA
24

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For more information see the Wikipedia article: [Polychromatic light](#)

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Revision #2

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